CLAIMS

1. A polycarbonate decomposition method comprising decomposing a polycarbonate with water in a supercritical or subcritical state.

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- 2. The decomposition method according to claim 1, wherein the polycarbonate is a polycarbonate contained in a thermoplastic composition containing the polycarbonate.
- 3. The decomposition method according to claim 1, wherein the polycarbonate is an aromatic polycarbonate and its decomposed product is an aromatic dihydroxy compound.
- 15 4. The decomposition method according to claim 3, wherein the aromatic polycarbonate comprises a recurring unit represented by the following formula (1):

$$\begin{array}{c|c}
 & R^1 \\
 & R^3 \\
 & R^4 \\
 & R^4
\end{array}$$

$$\begin{array}{c|c}
 & \cdots & (1) \\
 & R^3 \\
 & R^4
\end{array}$$

wherein R¹, R², R³ and R⁴ are each independently a hydrogen atom, alkyl group having 1 to 10 carbon atoms, cycloalkyl group having 6 to 10 carbon atoms, aryl group having 6 to 10 carbon atoms, aralkyl group having 7 to 10 carbon atoms or halogen atom, W is a single bond, alkylene group having 1 to 10 carbon atoms, alkylidene group having 2 to 10 carbon atoms, cycloalkylene group having 6 to 10 carbon atoms, cycloalkylidene group having 6 to 10 carbon atoms, alkylene-arylene-alkylene group having 8 to 15 carbon atoms, oxygen atom, sulfur atom, sulfoxide group or sulfone group, and the decomposed product is an aromatic dihydroxy compound represented by the following formula (2):

wherein R^1 , R^2 , R^3 , R^4 and W are as defined in the above formula (1).

- 5 5. The decomposition method according to claim 3, wherein the aromatic dihydroxy compound is recovered by crystallization.
- 6. The decomposition method according to claim 1, wherein the ion product Kw of water in a supercritical or subcritical state is 10⁻¹⁵ mol²/kg² or less.
- 7. The decomposition method according to claim 1, wherein the dielectric constant of water in a supercritical or subcritical state is 10 or less.
 - 8. The decomposition method according to claim 1, wherein decomposition is carried out at a temperature of 374 to 500°C.
- 9. The decomposition method according to claim 1, wherein decomposition is carried out at a pressure of 18 to 40 MPa.
- 10. An aromatic dihydroxy compound aqueous solution containing 1 wt% or less of an aromatic dihydroxy compound dissolved in water at a temperature of 10 to 100°C and a pressure of 0.1 to 10 MPa.